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Results for the cooling curve of paraffin wax: Graph showing the cooling curve of paraffin wax: As the time increases, the temperature of the paraffin wax decreases. Therefore, in conclusion, the paraffin wax in its liquid state changes to solid state as more time goes by due to the decrease in temperature.

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Chemistry 1 Experiment #1: The Cooling Curve of Stearic Acid INTRODUCTION Matter around us exists in three common states-solid, liquid, and gas. Matter can change from one state (or phase, as it is sometimes called) to another. Ice, for example, is the solid state of H₂O. Add

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Materials: test tube, naphthalene, thermometer, hot water, burner, stand, clamp. Be careful as naphthalene vapour is poisonous: you can wear a breathing mask to reduce the risk of you inhaling the vapour Method. Fill a test tube with naphthalene until it is about a quarter full; Place the test tube in hot water to allow the naphthalene to melt

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Cooling curves are the opposite. They show how the temperature changes as a substance is cooled down. Just like heating curves, cooling curves have horizontal flat parts where the state changes from gas to liquid, or from liquid to solid. You are likely to have used salol or stearic acid in a school practical lesson to make your own cooling curve.

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For my chemistry lab, I heated stearic acid at a constant rate until it transformed into a liquid and then reversed the process by cooling to produce a solid again. In this experiment, 1) What is going on at the molecular level as the liquid stearic acid cools and freezes?? 2) How would an increase in the amount of stearic acid used affect the shape of the heating and cooling curves?

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- Yes - change in potential energy. - Phase changes are going on, and as it is melting, it indicates that there is a change in potential energy - known as there is a change in position of atoms. - The temperature remains the same at the plateau; therefore, there is no change in kinetic energy. Click again to see term ?

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